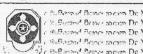
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Final Term Test - Grade 13 - July 2019

Index Number

#### **BIOLOGY I**

Time: 3 hours

Answer all questions.

- 1. Which is not a monosaccharide,
  - 1) Glucose

3) RuBP

5) Fructose

2) Maltose

- 4) Ribose
- 2. Select the correct monomer of the given polymers,
  - 1) Inuline
- Fructose
- 2) Hemicellulose Glucosamine
- 3) Chitin
- Pentose
- 4) Pectin
- Glucose
- 5) Glycogen
- Galacturonic acid
- 3. A nucleotide that act as an oxidative agent,
  - I) NADH

3) GTP

5) AT.

2) ATP

- 4) NADP+
- 4. False statement about cell membrane,
  - 1) Cholesterol molecules provide flexibility
  - 2) It is selectively permeable
  - 3) It is about 7 mm thick
  - 4) It is compared to the fluid mosaic model.
  - 5) Some proteins have short branching carbohydrates.
- 5. Correct statement about microscope.
  - 1) Visible light is passed back and forth through the specimen in light microscope.
  - 2) Electron beam is used in electron microscope.
  - 3) Resolution power of electron microscope is 0.2mm.
  - 4) Image is directly observed in electron microscopes.
  - 5) Dyes are used to stain the specimen in SEM.
- 6. Correct statement about flagella.
  - 1) Made of microtubes.
  - 2) Basal bodies has 9 + 2 arrangement.
  - 3) Covered by cytoskeleton.
  - 4) Consists of microbodies.
  - 5) It is a short cylindrical structure.
- 7. Which is not a significance of mitosis.
  - 1) Asexual reproduction.
  - 2) Growth and development.
  - 3) Maintains the genetic stability.
  - 4) Cell repair and replacement.



5) Produce new genetic variation. 8. Select the correct order of following steps in meiosis a) Haploid four daughter nuclei are formed. b) Kinetochore of each homologue attach to microtubules from one pole. c) Sister chromatids of each chromosome move toward opposite poles d) Homologues chromosomes arranged randomly at metaphase. 5) bdac 3) bdca 1) abcd 4) dbca 2) dcba 9. select the correct order of evolution of biodiversity. a) Development of multicellular Eukaryotes b) Accumulation of RNA into lipid bound vesicles and formed in to protocell c) Development of early photosynthetic organism. d) Origin of ancestors of arthropods and chordates. e) Evolution of sponges. 3) cbead 1) bcaed 2) cbaed 4) bcade 10. Which is not a common feature of Basidiomycota 1) Production of exogenous basidiospores 2) Possess dikaryotic filaments 3) Hyphae are aseptate. 4) They are decomposers. 5) They produce basidiocarp in sexual reproduction. 11. False statement about invertebrate's circulatory system. 1) In echinoderms circulatory system is reduced and closed without a heart. 2) Mollusca possess open circulatory system with haemocoel 3) Platyhelminthes possess an open circulation with out a heart. 4) Annelids possess close circulation with many hearts. 5) Nematodes are absence of circulatory system 12. Correct statement about photosynthesis. 1) Chlorophyll a is more effective for blue and red light. 2) Excited carotenoids boost their electrons to a higher level and become positively charged. 3) ATP and NADPH are produced in cyclic photophosphorylation. 4) 1, 3 – bisphosphoglycerate oxidic to G3 P. 5) Rubisco enzyme is prominently found in thylakoid membranes. 13. False statement about anaerobic respiration. 1) Occurs in mature red blood cells. 2) Final electron acceptor of alcoholic fermentation is acetaldehyde. 3) CO<sub>2</sub> is produce during alcoholic fermentation. 4) ATP is producing when acetaldehyde oxidized in to ethanol. 5) Occurs in the absence of oxygen. 14. Which is not common to epithelial tissues? 1) Ability to contract 2) Covers free surfaces of the body 3) Basal surfaces of cells attached to the basement mom 4) Absence of blood vessels

5) Apical surfaces of cells remain free

- 15. Incorrect statement about cells of ground tissue system of plant shoot.
  - 1) Parenchyma cells are dead cells at maturity.
  - 2) These cells are specialized for storage and photosynthesis.
  - 3) Collenchyma cells are generally elongated
  - 4) Sclerite cell wall is thickened by lignin.
  - 5) Parenchyma cells possess large central vacuole.
- 16. Correct statement about secondary growth in plants
  - 1) Occurs only in dicot plants.
  - 2) It increases the height of the plant.
  - 3) Xylem vessels achieve their conducting ability due to secondary growth.
  - 4) Periderm is the tissues that are interior to cork cambium.
  - 5) Periderm is impermeable to water and gases.
- 17. Select the correct deficiency symptoms in plants.
  - 1) Fe - roots stunted and excessively branched.
  - 2) Cu - Light purple colour through our young leaves.
  - 3) Mg - Crinkled leaves.
  - Crinkling of young leaves. 4) Ca
  - Chlorosis between veins.
- 18. Select the correct example given for the respiratory structures
  - 1) Body surface Scorpions
  - 2) External gills Aves
  - 3) Internal grills Marien annelids
  - 4) Skin
- Amphibia
- 5) Book lungs - Some insects
- 19. Change that does not occur during ventilation of the lungs.
  - 1) Pressure gradient is created between the atmosphere and lungs.
  - 2) During inhalation external intercoastal muscles contract.
  - 3) Exhalation is a passive process.
  - 4) During inhalation ribcage move up and out.
  - 5) Volume of lungs increase an visceral pleural relaxes.
- 20. Select the blood group of the donor when O(-) recipient is transfused.
  - 1) AB(-)

3) O(-)

2) AB(+)

4) O(+)

- 21. Which is not a constituent in blood plasma
  - 1) Hemoglobin
  - 2) Respiratory gases
  - 3) Albumin
  - 4) Antibodies
  - 5) Fibrinogen
- 22. False statement about human blood tissue.
  - 1) Agglutinogen is found on the surface of Red blood cells.
  - 2) Platelets are derived from epithelial cells of spleen.
  - 3) PH of human blood is around 7.4
  - 4) Leukocytes develop immune responses in an infection
  - 5) Erythrocytes are biconcave disk like cells.
- 23. A substance that does not secreted in to tubular filtrate in urine formation
  - 1) H+

2) K<sup>+</sup>

3) Creatinine



4) Na+

5) Penicillin

#### 24. Which is not a part of kidney tubule of human

- 1) Browman's capsule
- 2) Glomerulus
- 3) Loop of Henle
- 4) Distal convoluted tubule
- 5) Proximal convoluted tubule

#### 25. False statement about excretory structures in animals

- 1) Sweat glands are coiled tubular glands.
- 2) Nephron is the structural unit in human kidney
- 3) Malpighian tubules open in to the digestive tract in mollusks
- 4) Nephridia are multicellular tubular structures
- 5) Salt glands occur in pairs

#### 26. False statement about acquired immunity.

- 1) It is against a broad range of pathogens
- 2) Effector cells provide primary immune responses
- 3) Antibodies can neutralize and inactivate specific toxins
- 4) It is characterized by immunological memory
- 5) Acquired immune responses are mediated by lymphocytes.

#### 27. Which is not a sign of inflammatory responses

- 1) Red ness
- 2) Heat
- 3) Swelling
- 4) Production of helper T cells
- 5) Pain

#### 28. Select the false statement about fetal changes

- 1) Heart begins to beat by the 4th week.
- 2) Fetus grows to about 30cm in length at the end of first trimester.
- 3) Embryo is said to be fetus by the 8th week.
- 4) Fetal activity decreases at the end of third trimester.
- 5) Rudiments of all major organ systems appear in the first trimester.

#### 29. False statement about asexual reproduction

- 1) Mutations leads to new genetic variation in gametogenesis.
- 2) Rapid multiplication of organisms.
- 3) Asexual reproduction is more advantageous in stable environments.
- 4) Need only a single parent.
- 5) Asexual reproduction relies on mitotic cell division.

#### 30. Correct statement about the growth of zygote.

- 1) Cleavage of zygote begins in the urates
- 2) Cleavage of zygote take place about 6 hours after fertilization
- 3) Morula gets nutrition by endometrial secretions
- 4) Germ layers are formed before implantation.
- 5) Amnion involve in formation of placenta.

#### 31. Correct statement about sperm to genesis

- 1) Spermatogenesis occurs in epididymis
- 2) It starts at embryonic stage
- 3) Primary spermatocytes are produced by meiotic division of spermatogonia.
- 4) It takes 4 weeks to form mature sperm from spermatogonium

5) Once spermatogenesis process is done, sperms are release into the fluid - filled cavity. 32. False statement about micro - organism 1) Mollicutes are prokaryotic organisms. 2) Viruses get into the host with; they multiply and cause infections. 3) Unicellular protists are pleomorphic 4) Mutualistic fungi are found in lichens and mycorrhizae 5) Nostoc is free-living form of cyanobacteria that able fix atmospheric N2 33. Correct statement about food spoilage 1) Accumilation of toxins is a chemical change 2) Rancidity is caused by proteolytic bacteria 3) Food intoxication is caused by Shigella. 4) Aflatoxins are produced by Aspergillus flavus 5) Relative humidity does not affect on food spoilage. 34. Select the correct order in translation process. a) Amaino acids are added to the C-terminus of the growing polypeptide chain. b) Release of the t RNA in the P site c) t RNA carries methionine d) A sit is then ligned with the next codon. e) Small subunit of ribosome binds to m RNA 1) abdce 3) ecdab 5) ecabd 2) eacbd 4) abcde 35. Which is not an application of DNA sequencing. 1) To reveal evolutionary relationships within and among species. 2) Diagnosis of cancer 3) To determine the person on carrier or not regarding family inherit disease. 4) To understand the functions of DNA 5) To make recombinant vectors 36. False statement about human upper limb 1) Carpel bones are arranged in two rows 2) Radius articulate with three carpel bones at its distal end to form wrist joint 3) Opposable thumb permits precision grip 4) Ulna articulate with the distal end of humerus at elbow joint 5) Humerus forms a complete ball socket joint in glenoid cavity 37. Which is not a function of human skeletal system 1) Production of blood cells 2) Support 3) Movement 4) Protection 5) Transportation 38. Select the false statement about human genetic disorders 1) In males x - linked recessive disorders are expressed only at their homozygous genotype 2) Red – green colour blindness is caused by x – linked recessive disorder 3) Haemophilia is caused by x - linked recessive disorder 4) Cystic fibrosis is an autosomal recessive disorder 5) Sicke cell disease is caused by single gene mutation.

Answer the questions 39 and 40 according to the given description. A man of A blood group married to a woman with B blood group. Their son's blood group is O and daughter is AB 39. What are the gene types of parents for blood group? 1) IA IA x IB IB 3) ii x IA IB IAi x IBIB 2) IAi x IBi 4) IAIA x ii 40. What is the probability of daughter with AB blood group? 1) 1/8 3) 1/2 5) 1/16 2) 3/8 4) 1/4 Answer questions 41 - 50 using the given instruction ABD ACD AB CD Other correct correct correct correct combination 41. Paired bones found in human cranium A. Sphenoid C. Parietal bone E. Frontal B. Mandible D. Temporal bone 42. Select the incompatible relationship between the micro - organism and the human disease A. Typhoid - Salmonella typhi B. Bacterial meningitis - Streptococcus pheumonige C. Tuberculosis - Mycobacterium tuberculosis D. Pneumonia - Streptococcus pneumoniae E. Food poisioning - Staphylococcus aureus 43. Which are is not a macro - molecule A. Keratin C. Phospholipid E. Glycogen B. Ribose D. Collagen 44. Which is nota a main part of human brain A. Spinal cord C. Thalamus E. Pituitary B. Spinal nerves D. Medulla oblongata 45. False statement(s) about human skin A. Dermis composed of keratinized epithelium B. Blood capillaries extend in to epidermis C. Collagen fibers provide tensile strength D. Melanocytes found in deep layers of dermis E. Skin is the largest organ in the human body 46. Hormones released by anterior pituitary A. Oxytocin C. Prolactin E. CRH B. ADH D. AcTH 47. Coastal ecosystems found in Sri Lanka A. Sand Dunes C. Savana E. Patana B. Sea shore D. Villu 48. Select the incorrect relationship according to IUCN publishes A. Extinct - Dodo B. Endanger - Elephant C. Vulnerable - Marbled rock frog D. Critically endangered - Jungle squirrel

- Thilapla

E. Exotic species

- . Effects of global warming
  - A. Degradation of coral reefs
  - B. Sea level rising
  - C. Depletion of ozone layer
  - D. Extreme water events
  - E. Photochemical smog
- ). Which is / are contributary factors for desertification
  - A. Acid rains
  - B. Over exploitation of soil
  - C. Uncontrolled mining
  - D. Excessive use of agro chemicals
  - i. Emission of green house gases.

## DE MAZENOD COLLEGE – KANDANA THIRD TERM TEST – 2019 JULY BIOLOGY – GRADE 13

01.08.2019

3 HOURS

	Part A -	- Structured Essay	and the major processes than	TEN III i
Answer the qu	uestions using the given space.			
1. A)				•
i. Name the element	s that found as 96% in living m	natter.		
ii. Why water is imp	ortant to life?			
************************	***************************************	•••••		,
iii. Name a Nitrogen	containing polysaccharide.			
iv. Name a Structural	polysaccharide found in proka	aryotes.		
********************			,	•
vi. (a) What is meant	by resolution power of microse	cope?		
(b) What is the res	solution of light microscope?			
- in Position dies in	the ultra-structure of mitochor	Thing of the state	Resource College Kandana	M.II
(b) Wha	at are the functions of the abov	e organelle?	Vint is mount by radial trans ame the different types of ro	K.
B)				
i. What is mea	nt by respiratory quotient?			

ii. What are the major differences of lactic acid fermentation and Alcoholic fermentation?
50 - 86 - 10 .
SC - 86 - 10.
iii. What are the major processes that take place in CO2 fixation in C3 plants?
Apewer the guestions using the piven space.
Alternative at the second of t
Name of the control o
iv. What are the factors that affect on the rate of photosynthesis?
Why water is important to like the control of the c
***************************************
Name a Nitrogen containing polysaccharide
v. What are allosteric regulators?
vi. What are the changes occur in 'S- 'phase of interphase in mitotic cell division.
(C)
i. Name the significances of mitosis
ii. Briefly explain what is meant by crossing over.
2 4
A)     i. What is meant by radial transport?
ii. Name the different types of routes used in the radial transport.
the table the different types of routes used in the radial transport.
••••••

#### 1.08.2019 Grade - 13 Biology

	iii. What is meant by nutrition in plants?
	iv. What are the needs of nutrition in plants?
	. Will also the second of the plants of
	v. Why essential elements are needed for plants?
	vi. Name deficiency symptoms of Ca in plants.
	Numbrooms are do remeded readounce and others for an
V	ii. Name the source and the form of absorption of P into plants.
	Source: -
	Form:
vi	ii. Name the soil microbial interaction in effective uptake of P into plants.
	A) Maine the effetent components of peripheral nervous syntem
3)	
	i. Name the key steps in the Nitrogen cycle.
	ere or car read and our acceptant and enterprise
	Si what are the sandications of miner armonisms in action/ture?
	n. what are the applications of infero-organisms in agriculturer
	iii. Name the indicator organism used to test the quality of drinking water.
	iv. State the characteristic features of the above mentioned organism.
	v. Name two water born diseases



	vi. What are the adverse effects of discharging large amount of waste water into natural water bodies?	
p 14		
(C)	i. What is the functional unit of liver?	
	***************************************	
	ii. What are the functions of liver related to digestion?	
	iii. (a) Name the hormones released by the duodenum.	
	(b) What are the functions of the above hormones.	
	1979-1	
	-3009	
3. A)	i. (a) Name the efferent components of peripheral nervous system	
	a	
	b	
	b) Write the function of each of the above-mentioned components.	
	2	
	b	
	ii. Name the functions of human cerebellum	
	in reduce the functions of numan cerebellum	
	RESIDENT	
	iii. What are the factors required to maintain the resting membrane potential?	
	iv. What are the phases of action potential?	
74,500		
	v. Name the different types of photoreceptors in human eye and state the visual pigment in those receptors	
	P.B. and the strong of the str	

Plac 80-1 El - aborde Upolaid

iii. Red-green colour blindness is caused by sex linked recessive allele. In a family of normal parent's se is colour blind and their daughter is normal for colour blindness. write the genotypes of family (Dominant allele = C recessive allele = c) Mother ..... Father ..... Son ...... Daughter ..... iv. What is meant by mutation? v. What are the different types of gene mutations? B) i. Name three genetic disorders that occur due to chromosomal mutations. ii. State major steps and techniques used to make a recombinant vector? iii. What is meant by cloning site? iv. Name an international agreement established in order to safeguard consumers, society and environment due to potential risks of GMO s? v. What is the objective of the above mentioned agreement? vi. What are the functions of DNA polymerase in DNA replication? 

vii. What are the	three main steps in elongation of translation?
(C) i. State 5 char	racteristic features of the vegetation in desert biome in tropics?
1	
2	
3	
4	
5	
	All the state of t
ii. Name the valu	ues of Biodiversity?
######################################	SHET 1 YEAR COMM.
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	Vice adoption and taward.
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iii. Write two hu	uman activates leads to descrification?

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Index Number .....

#### **BIOLOGY II**

Time: 3 hours

Part B – Essay
 Answer four questions only.

#### PART B - ESSAY

- 1. Explain the process of osmoregulation in human body.
- 2. Briefly explain the lung ventilation mechanism and its regulation
- 3. i. Explain the vitality of water in living matter with respect to its major properties.
  - ii. Explain the stomatal transpiration in higher plants.
- 4. i. Briefly explain the salar astructure of chloroplast in eukaryotic cell.
  - ii. Explain the C3 cyclic pathway of CO2 fixation in photosynthesis.
- 5 i. Explain sterilization techniques used in laboratory preparation of microbial culture
  - ii Briefly explain the physical and chemical changes that take place in microbial food spoilage.
  - 6. Write short notes.
    - i. Cardiac conduction system.
    - ii. Acid rains
    - iii. Gross structure of human ovary

#### Part B - Essay

#### Answer All Questions

- 5. (a) (i) Deduce the relationship between the relative molecular mass (M) and the density(d) of a particular gas by using the ideal gas equation.
  - (ii) A mixture of gases contains 75% of N<sub>2</sub> gas and 25% of O<sub>2</sub> gas by volume. The pressure of the mixture of gases is 1.00 x 10<sup>5</sup> Pa and the temperature is 300 K. Assuming ideal behavior, calculate the following.
    - a. The partial pressure of  $O_2$  in this mixture of gases
    - b. The relative molecular masses relevant for this mixture of gases. (the relative atomic masses of N and O are 14.0 and 16.0 respectively)
    - c. The density of the mixture of gases.

(iii) Various mixtures consisting with finely powdered Al, Mg and Zn were prepared, each weighing 100 g. Masses of different metals used for the preparation were given below.

Mixture	M	ere given below.			
a talka da ka	Al	Zn	Mg	Total mass/(g)	
Р	21.6	13.0	65.1	211 ( 316809)	
Q	27.0	52.0	05.4	100.0	
R	32.4	65.0	21.0	100.0	
9.1%-		03.0	2.6	100.0	

(Al=27, Mg=24, Zn=65)

However the labeling of the samples was found to be inaccurate. Using balanced chemical equations and proper calculation, how would you identify the samples separately?

- (iv) Speeds of six gaseous molecules of an ideal gas measured at -250 °C were given as, 2.0 m s<sup>-1</sup>, 2.2 m s<sup>-1</sup>, 2.6 m s<sup>-1</sup>, 2.7 m s<sup>-1</sup>, 3.3 m s<sup>-1</sup> and 3.5 m s<sup>-1</sup> respectively. Using these values calculate the following,
  - a. Average speed of a gas molecule
  - b. Root mean square speed

- c. Molar mass of the gaseous species
- d. Density of the gas at 10 Pa pressure
- 6. (a) Consider the chemical reaction,  $2Na_{(s)} + \frac{1}{2}O_{2(g)}$  Na<sub>2</sub>O<sub>(s)</sub>

Some of the thermo chemical data given relevant to the above reaction are given below.(at 25°C.)

Atomization enthalpy of  $Na_{(s)}$  = 107 kJ mol<sup>-1</sup> First ionization enthalpy of  $Na_{(g)}$  = 494 kJ mol<sup>-1</sup> Atomization enthalpy of  $O_{2(g)}$  = 249 kJ mol<sup>-1</sup> First electron affinity of  $O_{(g)}$  = -149 kJ mol<sup>-1</sup> Second electron affinity of  $O_{(g)}$  = 798 kJ mol<sup>-1</sup> Lattice enthalpy of  $Na_2O_{(s)}$  = 2522 kJ mol<sup>-1</sup>



- I, Write chemical equations to express above enthalpy changes.
- Calculate the enthalpy change of formation of Na<sub>2</sub>O<sub>(s)</sub> by constructing an appropriate 11. enthalpy level diagram.
- Standard enthalpy change of hydration of  ${\rm Li^+_{(g)}}$  ,  ${\rm Na^+_{(g)}}$  and  ${\rm Cl^-_{(g)}}$  are -499 kJ mol<sup>-1</sup> , -390 6) kJ mol<sup>-1</sup> and -381 kJ mol<sup>-1</sup> respectively. And standard enthalpy change of lattice of LiCl<sub>(s)</sub> and NaCl<sub>(s)</sub> are -864 kJ mol<sup>-1</sup> and -780 kJ mol<sup>-1</sup>.
- Calculate the standard enthalpy change of dissolution of LiCl(s) and NaCl(s)
- II. State, whether the temperature of the system increase or decrease, when LiCl(s) is dissolving in water.
- c) Calculate the enthalpy of formation of NaCl(s) from the following data

Reaction	owing data
$NaOH_{(aq)} + HCl_{(aq)} \rightarrow NaCl_{(aq)} + H_2O_{(t)}$	ΔH <sup>0</sup> <sub>298</sub> / kJ mol <sup>-1</sup>
$HCl_{(g)} + aq \rightarrow HCl_{(aq)}$	-57.3
$H_{2(g)} + \frac{1}{2} O_{2(g)} \rightarrow H_2O_{(1)}$	-71.9
$\frac{1}{2}$ $H_{2(g)} + \frac{1}{2}$ $Cl_{2(g)} \rightarrow HCl_{(g)}$ and an install assume as legalities and the second	-285.9
$Na_{(s)} + \frac{1}{2} O_{2(g)} + \frac{1}{2} H_{2(g)} \rightarrow NaOH_{(aq)}$	92.3
$NaCl_{(s)} + water \rightarrow NaCl_{(aq)}$	-425.6
molecules of an ideal gas measured at (ps) of the given as 2.0	+3.9

- (a)" L" is an element that belongs to the p block of the periodic table. In the group that L belongs, there are metals, non metals as well as metalloids. The first member of the group is distinctive as it has a tendency to form a large amount of compounds. L is the second member of this group.
  - Identify L
  - Write the complete electronic configuration of L ii. iii.
  - Write the chemical formula and the colour of the most stable oxide of L.
  - L is very important in the manufacture of integrated circuits and transistors. State iv.
  - (b) Consider the oxidation of  $SnC_2O_4$  by  $KMnO_4$  in the dil.  $H_2SO_4$  medium.
    - Write the reduction half reaction ii.
    - Write the oxidation half reaction iii.
    - Give the balanced redox reaction

- (c) You are supplied with a mixture consisting of barium carbonate, dolomite, potassium carbonate and silicon dioxide. Explain how you would attempt to determine quantitatively each of the constituents present in this mixture.
- (d) When 20.0 g of a white solid X is heated, 4.4 g of an acidic gas A and 1.8 g of a neutral gas B are evolved, leaving behind a solid residue of mass 13.8 g. A turns lime water milky and B condenses into a liquid which changed anhydrous cupric sulphate blue. The aqueous solution Y is alkaline to litmus and gives 19.7g of white precipitate Z with barium chloride solution. Z gives carbon dioxide with an acid. Identify A, B, X, Y and Z. (C= 12, H= 1, O = 16, Ba= 137)
- (a) State all possible reactions (balanced) which can be occurred when a piece of Na is exposed
  to air.
  - (b) How would you distinguish between the below mentioned pairs?
    - i. Li<sub>2</sub>CO<sub>3</sub> and Na<sub>2</sub>CO<sub>3</sub>
    - ii. Na<sub>2</sub>CO<sub>3</sub> and MgCO<sub>3</sub>
    - iii. LiNO3 and NaNO3
    - iv. KNO<sub>3</sub> and Ba(NO<sub>3</sub>)<sub>2</sub>
    - v. Na<sub>2</sub>CO<sub>3</sub> and NaHCO<sub>3</sub>
    - vi. LiOH and NaOH
  - (c) What will happen to the water solubility of below mentioned species formed by alkaline earth metals down the group?
    - i. Carbonates
    - ii. Sulphates
    - iii. Oxalates
    - iv. Hydroxides
    - v. Chromates
  - (d) Assume that you have provided a mixture containing KNO<sub>3</sub> and K<sub>2</sub>CrO<sub>4</sub>. You are provided Ba(NO<sub>3</sub>)<sub>2</sub> and CH<sub>3</sub>COOH as the only chemical reagents. How would you confirm the presence of KNO<sub>3</sub> using the given reagents. (Heating is allowed)
  - (e) X is an element of the main group of the periodic table. Nitrate of the element X is soluble in water forming a solution Y. White precipitate Z can be obtained due to the drop wise addition of dil. NaOH to solution Y. Z is soluble in excess NaOH forming a colourless solution P. Z can be re-emerged due to the drop wise addition of dil. HCl to solution P. Z is soluble in excess dil. HCl forming solution Y. Identify X,Y,Z and P.
  - (f) A mixture of zinc and aluminum weighing 1.67 g was completely dissolved in dilute sulphuric acid and evolved 1.69 L of hydrogen gas measured at 273 K and 1 atm pressure. What was the weight of aluminum in the original mixture? (Assume the ideal behavior of hydrogen gas, Al= 27, Zn = 65)

Li	Be	W. Williams			d stati	CHIE II	Trius s		ens be			13	14	15	bg.	ta ta	
Na Na	Mg	301		reni	ODIC	TABLE	OF T	HE ELI	EMENT	S		B	C	N National National	white transmission and the	F	1
.n. a. 15	20	21	7 29	5	Q	7	a	9	1 10			Al	SI	P	S	CI	7
M	Ca	Sc	TI	V	Čr.	Mn.	Fe Fe	27	20	14	7 30	Mar 31	skeen seed	Manhates M.K.	HE OF HE PER	dégray rate auti	1
37	30	38	State of the state	99.00	Droffson Mar	Stangerage.	- mip	Co	NI	Cu	Zn	Ga	Ge	As	Se 34	14	
Rb	Sr	Y	Zr	Nb	Mo Mo	43	44	45	48	47	de par	Section 1	Softmany QL:	A Harris	MARIAGAS.	Br	1
525	HTTP:	HERIT HERIT	Embedge.	feetback.	- Propriestation	To	Ru	Rh	Pd	Ag	Cd	lin lin	Sn	Sb	62	50	· ·
Cs	Sa	67.71	Hf	7.3	14 74	FB	76	77	Section of the sectio	70	ricedistroph tile	Attheb	W 191	serionizing.	TO HOUSE	entair.	X
ins.	Burkeys He o	tentamorina	Politically:	Ta	wester	Re	Os	lr Ir	P	Au	Hg	11	BQ.	RANGE CONTRACTOR	BA	65 (84	100
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		to B	Ce	Pr	No	Pm	Sm	Eu	61	65	68	67 7	- EN T	50 7	76 1		
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		Ac	Th	Pa	92 U	93 May	94	91	94	97	113.1 665	161 h	HOA .	100.1	Statorie 1214	MAGEN F	
	Ĺ	MARINES.	1925	tackerquires	CHE SALE	Np	Philippine I	Am	Cm	Bk	Cf	Es	Fm	Wd.	No	193 Lr	

Dec(NO2); and CH5COOH as the only chemical reagents. How would you confirm the previous of KNO3 using the given reagents. (Heating is allowed), and or specied that the confirm the presence of KNO3 using the given reagents. (Heating is allowed), and or specied that the

ceramoid?

(c) X is an element of the natary your of the periodic house of the element X is reliable to water forming a submitted previously 2 sear bit construct due to the drop wise attition of distribution to softwink Y Z is soluble in excess itsOH forming a coloublest section P Z can be re-emerged due to the drop wise addition of distributions P Z is soluble in excess dil. HCl forming solution Y attentive XCM and P is a control table of the solution Y attentive XCM and P is a control table of the solution.

A mixture of zinc and aluminute verglang Lot, governormed distributed in communities and and contractly for 1 or hydrogen gas measured as 2.73 K and 1 arm contract.

What was the weight of aluminum in the original mixture? (Assume the ideal sehawter of the force gas,  $\Lambda = 27$ ,  $Z_R = 65$ )

The dropen gas,  $\Lambda = 27$ ,  $Z_R = 65$ )

The dropen gas are a first the set of the force of the force gas are a first to the force of the first to the firs